

Al-Cr (Aluminum-Carbon-Chromium)

V. Raghavan

Schuster et al. [1980Sch] presented two isothermal sections at 1000 and 800 °C for this system. Recently, [2006Hal] determined the thermodynamic properties of the ternary compound Cr₂AlC and computed a liquidus projection and two isothermal sections at 1400 and 1000 °C.

$\beta\text{Cr}_4\text{Al}_9$, $\gamma\text{Cr}_4\text{Al}_9$, $\alpha\text{Cr}_5\text{Al}_8$, and $\beta\text{Cr}_5\text{Al}_8$, with no clear phase boundaries between them [2000Mah]. The C-Cr phase diagram [1990Ven] depicts three well-established compounds Cr₂₃C₆ (D_{84} -type cubic), Cr₇C₃ (D_{10_1} -type orthorhombic), and Cr₃C₂ ($D_{5_{10}}$ -type orthorhombic).

Binary Systems

In the Al-C system [1991Har], the stoichiometric compound Al₄C₃ (D_{7_1} -type rhombohedral) is present. The Al-Cr phase diagram by [2000Mah] includes a thermodynamic assessment and depicts a number of intermediate phases with significant ranges of homogeneity: CrAl₇ (V₇Al₄₅-type monoclinic), Cr₂Al₁₁ (CrAl₅-type monoclinic), CrAl₄ (monoclinic), Cr₂Al (MoSi₂-type tetragonal), and an unconfirmed low-temperature phase X at 75 at.% Cr. Between 30 and 41 at.% Cr, five phases have been reported: $\alpha\text{Cr}_4\text{Al}_9$,

Ternary Phase Equilibria

[1980Sch] determined the crystal structure of the ternary compound Cr₂AlC (denoted τ here; hexagonal, $P6_3/mmc$, 8 atoms/cell, $a = 0.2866$ nm and $c = 1.282$ nm). Employing ab initio calculations, [2006Hal] estimated the energy of formation of Cr₂AlC. The incongruent melting temperature determined from the not-so-precise DTA measurement was between 1460 and 1495 °C and agrees satisfactorily with the calculated value of 1498 °C. [2006Hal] accepted the binary descriptions of [1995Gro] (Al-C), [1992Lee] (Cr-C)

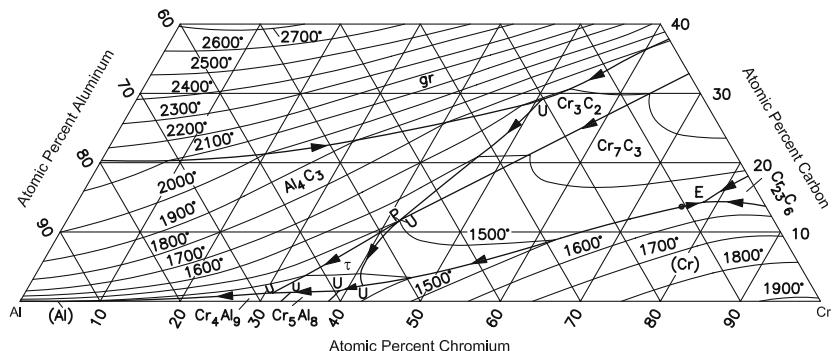


Fig. 1 Al-Cr computed liquidus projection [2006Hal]

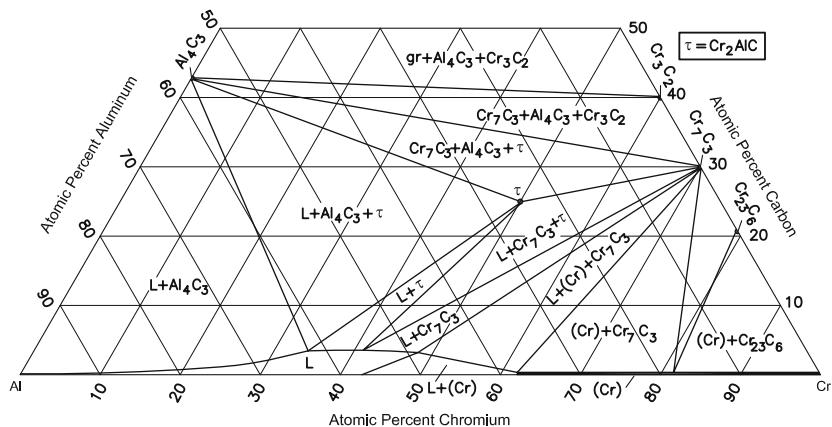


Fig. 2 Al-Cr computed isothermal section at 1400 °C [2006Hal]. Narrow two-phase regions are omitted

Section II: Phase Diagram Evaluations

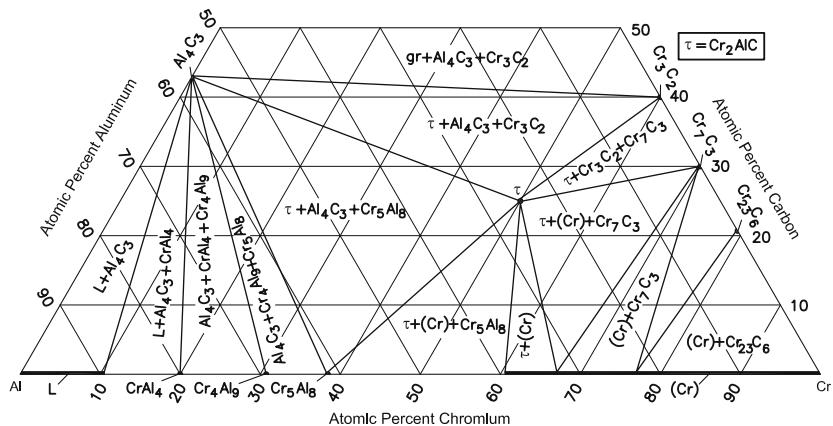


Fig. 3 Al-Cr computed isothermal section at 1000 °C [2006Hal]. Narrow two-phase regions are omitted

and [1998Sau] (Al-Cr). All compounds including the Al-Cr phases were treated as stoichiometric.

A liquidus projection and two isothermal sections at 1400 and 1000 °C were computed by [2006Hal], see Fig. 1-3. The invariant reactions on the liquidus surface (Fig. 1), their temperatures and compositions were listed. The primary crystallization regions of (Al), CrAl_7 , $\text{Cr}_2\text{Al}_{11}$, and CrAl_4 are very close to the Al-Cr side in Fig. 1. In the isothermal sections (Fig. 2 and 3), the three-phase field of $(\text{Al}_4\text{C}_3 + \text{Cr}_3\text{C}_2 + \text{gr})$ extends up to the graphite corner. The triangulation seen at 1000 °C in Fig. 3 agrees with that of [1980Sch]. The tie line between τ and graphite reported by [1980Sch] at 800 °C could not be reproduced in the calculations.

References

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